

## COUMARINS OF *Persica vulgaris*

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Leaves and year-old runners of common peach contain several phenolic compounds [1]. However, the coumarin composition of this plant has not been reported. Results from a study of coumarins obtained from various parts of common peach (*Persica vulgaris* L., Rosaceae) are presented here.

We used leaves, flowers, and branches of peach collected in summer 2001 in the Crimea. The raw material was ground to 2-3 mm, treated with ethanol (70%), and evaporated to an aqueous solution. Coumarins were extracted by  $\text{CHCl}_3$ . The  $\text{CHCl}_3$  fractions were chromatographed using  $\text{CHCl}_3$ — $\text{HCONH}_2$  and petroleum ether—formamide. Authentic coumarins were used as standards. The chemical composition was deduced based on five parallel determinations. The  $R_f$  values of spots before and after treatment with chromophoric reagents revealed at least six coumarin-type compounds upon visualization in UV and natural light. We found umbelliferone in all parts, flowers and branches more than leaves; and scopoletin, coumarin, and scopolin in flowers and branches.

Quantitative determination of the coumarins used photolorimetry after conversion of coumarins to the sodium salt of the corresponding coumarinic acids [5]. Leaves contain 2.34% coumarins; branches, 2.60%; flowers, 3.91%.

Coumarins were isolated from air-dried peach branches (2 kg) by grinding them and treating them with seven times the volume of ethanol (70%). The extract was evaporated to an aqueous solution, from which the precipitated lipophilic compounds were filtered off. The filtrate was treated with  $\text{CHCl}_3$ . The solvent was evaporated. The solid after evaporation of  $\text{CHCl}_3$  was placed on a polyamide column that was eluted with benzene,  $\text{C}_6\text{H}_6$ — $\text{CHCl}_3$  mixtures, and  $\text{CHCl}_3$ . The separation was monitored by paper chromatography and TLC using petroleum ether—formamide and  $\text{CHCl}_3$ — $\text{HCONH}_2$ .

Four coumarinic compounds were isolated:

**Compound 1**,  $\text{C}_{10}\text{H}_8\text{O}_4$ , mp 200-202°C ( $\text{C}_2\text{H}_5\text{OH}$ — $\text{CHCl}_3$ ), identified as scopoletin [2, 4];

**Compound 2**,  $\text{C}_9\text{H}_6\text{O}_3$ , mp 230-232°C (ethanol), identical to umbelliferone [4];

**Compound 3**,  $\text{C}_9\text{H}_6\text{O}_2$ , mp 67-68°C (ethanol), identical to coumarin [4];

**Compound 4**,  $\text{C}_9\text{H}_6\text{O}_3$ , mp 207-209°C (ethanol), identical to scopolin [4].

The compounds were identified by their physicochemical properties,  $R_f$  values in various solvent systems, UV and IR spectroscopy, and mixed melting points with authentic samples.

## REFERENCES

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